## The Stereochemistry of Melianone and Sapelin F: Correlation with Bourjotinolone A

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Summary Thermolysis of the epoxytriol (9) from melianone (4) gave 3-episapelin A (2), thus defining the stereochemistry of melianone (4) as 23-R,24-S; sapelin F (16) is shown also to have 23-R,24-S stereochemistry by correlation with melianone via 3-epimeliantriol (13).

Two of the modifications of the side chain of the protolimonoids<sup>1</sup> (highly oxygenated triterpenes related to tirucallol) are shown in bourjotinolone A (1)<sup>2</sup> and melianone (4).<sup>3</sup> Both (1) and (4) are, in principle, derivable from the epoxydiol (8), bourjotinolone A presumably being formed with inversion at C(24). However, while bourjotinolone A has been shown<sup>2</sup> to have 23-R,24-R stereochemistry and is hence derivable from an epoxydiol (8) which is 23-R,24-S,

OH

$$24$$
 $11$ , R = 0

 $12$ , R = H,  $\beta$ -OH

 $13$ , R = H,  $\alpha$ -OH

 $13$ , R = H,  $\alpha$ -OH

 $13$ , R = H,  $\alpha$ -OH

$$R^{20}$$
  $R = C_{22}H_{33}O$   $R^{2} = H$ ,  $R^{2} = C_{22}H_{33}O$   $R^{2} = H$   $R^{2} = C_{22}H_{33}O$   $R^{2} = H$   $R^{2} = C_{22}H_{33}O$   $R^{2} = H$ 

melianone (4) has been assigned the 23-S,24-R; configuration. On the assumption that (1) and (4) were derived from the same epoxydiol (8), and hence that the stereochemistry assigned to melianone (4) was wrong, we have sought a correlation between bourjotinolone A (1) and melianone (4).

Thermolysis at  $195^{\circ}$  in the absence of solvent of the epoxytriol (9), 3 obtained by sodium borohydride reduction of melianone (4), gave (2), \$\\$ the 3-epimer of sapelin A (3), 4 as the only isolable product (25% yield). (2) was made by sodium borohydride reduction of bourjotinolone A (1). This result demonstrates that melianone (4) is, in fact, 23-R,24-S, since the thermolysis must have involved inversion at C(24). Bevan et al. 5 had correctly assigned the 23-R configuration to turraeanthin (5), shown to differ from melianone (4) only at C(3), on the basis of the molecular rotation difference of the corresponding  $\gamma$ -lactone and its salt, but had made no assignment at C(24).

The assignment<sup>3</sup> of the 23-S,24-R stereochemistry to melianone (4) followed from the assignment of structure (10) for the product of treating melianone with HBr in acetic

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- ‡ In ref. 3, the deduced configurations were wrongly labelled 23-R,24-R.
- § The  $3\beta$ -alcohol (2) had been previously obtained by catalytic hydrogenation of bourjotinolone A (1), but was only characterised as its acetate.

acid, assuming that no inversion had taken place at C(20), (known to be 20-S).3 However, strong acids can reversibly give enols from acetals and hence C(20) must have changed from the less stable configuration in (11) to that in (12). The results of Lavie et al.3 do, however, define the threo relationship between C(23) and C(24) in melianone (4) and confirm that the conversion of (9) to (2) involves  $S_{N}2$  attack of the primary alcohol on the epoxide with inversion at C(24).

From Jamaican Melia azaderach, we have isolated, in addition to melianone (4) and melianodiol (15) (found3 in the same plant growing elsewhere), 3-epimelianol (6) and 3-epimeliantriol (13), m.p. 200-204°. The structures of (6) and (13) were deduced from their spectral data and confirmed by correlation with each other and with melianodiol (15). Thus, silver carbonate on celite oxidation of

3-epimeliantriol (13) gave, in fair yield, melianodiol (15), while hydrolysis of the epoxides in (6) and (4) with 1:1 0.1N sulphuric acid-tetrahydrofuran, conditions shown to give almost exclusive cleavage of epoxides via the more stable carbonium ion, gave cleanly 3-epimeliantriol (13) and melianodiol (15) respectively. Sodium borohydride reduction of (13) yielded sapelin F (16)? which is therefore 23-R,24-S. The three relationship between C(23) and C(24) in (16) had been previously deduced by the same arguments used for bourjotinolone C (17).2

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¶ The corresponding 3-acetates (7) and (14) have been isolated from Turraeanthus africanus, see J. G. St. C. Buchanan and T. G. Halsall, Chem. Comm., 1969, 48.

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